# (-)-5-endo,6-exo-Dihydroxycamphene 

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#### Abstract

C}_{10} \mathrm{H}_{16} \mathrm{O}_{2}, M_{r}=168 \cdot 24\), tetragonal, $, 4_{1}, a=$ 16.385 (2), $c=14.879$ (2) $\AA, V=3994.5 \AA^{3}, Z=16$ (two molecules per asymmetric unit), $D_{x}=1.12 \mathrm{Mg}$ $\mathrm{m}^{-3} . \mathrm{Cu}$ radiation, $R=0.047$ for 772 reflexions (diffractometer data). The norbornane skeleton is slightly distorted as a result of intramolecular steric repulsion. Molecules are linked in the crystal by a double helical system of hydrogen bonds.


Introduction. A simple synthetic route from camphor to 5,6 -dihydroxycamphenes has been developed with a view to assessing the potential phytohormone activity of these compounds (Allen, Lamb, Money \& Salisbury, 1979). The present study of the 5 -endo, 6 -exo isomer (I) was undertaken to provide structural data to assist in the interpretation of the biological function of these molecules.

(I)

A colourless needle crystal was used $\mathbf{( 0 . 2 \times 0 . 2 \times}$ 0.4 mm ), and unit-cell data were obtained by a leastsquares fit of $25 \sin \theta$ values measured on a GE XRD-6 diffractometer with $\mathrm{Cu} K \alpha$ radiation. Intensity data were measured on an Enraf-Nonius CAD-4F diffractometer, with $\mathrm{Cu} K \alpha$ (Ni filter), $\theta \leq 60^{\circ}, \omega / 2 \theta$ scan of $(0.80+0.15 \tan \theta)^{\circ}$ in $\omega$ extended $25 \%$ on either side for background measurement, variable scan speed, and check reflexion measurement every hour. Of 1557 reflexions with $\theta \leq 60^{\circ}, 772$ had intensities greater than $3 \sigma\left[\sigma^{2}(I)=S+B+(0.04 S)^{2}\right.$, where $S=$ scan, $B=$ time-corrected background count] (most of the weaker reflexions were at high angle). No absorption corrections were applied.

The structure was determined by direct methods, 226 reflexions with $|E| \geq 1.35$ being used; 11,10,1 was used to fix the origin, and eight pairs of sets of phases were developed by assigning two values to each of 220, $8,7,11,0,9,11$, and 022 to initiate tangent refinement. The $E$ map from the best solution gave positions for about two thirds of the atoms in the two independent
molecules, and the remainder were located on a subsequent $F_{o}$ map. A subsequent difference map gave positions for only about one half of the H atoms; from these, positions could be calculated for all the H atoms, including those involved in hydrogen bonding, with the assumption of ideal geometry. Refinement was completed by full-matrix least-squares methods, with minimization of $\sum w\left(\left|F_{o}\right|-\left|F_{c}\right|\right)^{2}$, with $w=1 / \sigma^{2}(F)$; anisotropic thermal parameters were used for C and O atoms, H atoms were included but not refined, and an extinction parameter was refined. Final $R$ was 0.047 for 772 reflexions ( 217 parameters). The final positional parameters are in Table 1.* The absolute configuration is established from the chemical preparation.

[^0]Table 1. Final positional parameters ( fractional, $\times 10^{4}$ ) with estimated standard deviations in parentheses

Unprimed atoms are in molecule 1, primed atoms in molecule 2.

|  | $x$ | $y$ | $z$ |
| :--- | :---: | :--- | :---: |
| $\mathrm{O}(5)$ | $1490(5)$ | $1809(6)$ | $2136(8)$ |
| $\mathrm{O}(6)$ | $1317(6)$ | $2544(6)$ | 0 |
| $\mathrm{C}(1)$ | $-64(8)$ | $2227(11)$ | $567(12)$ |
| $\mathrm{C}(2)$ | $-475(8)$ | $1562(11)$ | $1058(11)$ |
| $\mathrm{C}(3)$ | $-400(8)$ | $1784(9)$ | $2048(11)$ |
| $\mathrm{C}(4)$ | $163(8)$ | $2547(8)$ | $2002(10)$ |
| $\mathrm{C}(5)$ | $1026(8)$ | $2367(7)$ | $1624(10)$ |
| $\mathrm{C}(6)$ | $858(7)$ | $2066(7)$ | $640(9)$ |
| $\mathrm{C}(7)$ | $-186(10)$ | $2986(11)$ | $1165(13)$ |
| $\mathrm{C}(8)$ | $-782(13)$ | $886(16)$ | $708(15)$ |
| $\mathrm{C}(9)$ | $-69(9)$ | $1084(8)$ | $2641(12)$ |
| $\mathrm{C}(10)$ | $-1243(9)$ | $2040(12)$ | $2455(14)$ |
| $\mathrm{O}\left(5^{\prime}\right)$ | $3665(6)$ | $2683(5)$ | $3231(7)$ |
| $\mathrm{O}\left(6^{\prime}\right)$ | $3138(5)$ | $1578(5)$ | $1510(8)$ |
| $\mathrm{C}\left(1^{\prime}\right)$ | $4594(7)$ | $1938(8)$ | $1234(10)$ |
| $\mathrm{C}\left(2^{\prime}\right)$ | $5105(7)$ | $2693(9)$ | $1304(11)$ |
| $\mathrm{C}\left(3^{\prime}\right)$ | $5380(8)$ | $2758(8)$ | $2275(10)$ |
| $\mathrm{C}\left(4^{\prime}\right)$ | $4951(7)$ | $1988(7)$ | $2690(10)$ |
| $\mathrm{C}\left(5^{\prime}\right)$ | $4016(8)$ | $2045(8)$ | $2698(10)$ |
| $\mathrm{C}\left(6^{\prime}\right)$ | $3788(7)$ | $2148(7)$ | $1708(10)$ |
| $\mathrm{C}\left(7^{\prime}\right)$ | $5031(8)$ | $1369(8)$ | $1917(10)$ |
| $\mathrm{C}\left(8^{\prime}\right)$ | $5274(10)$ | $3195(11)$ | $631(13)$ |
| $\mathrm{C}\left(9^{\prime}\right)$ | $5132(11)$ | $3607(9)$ | $2684(13)$ |
| $\mathrm{C}\left(10^{\prime}\right)$ | $6325(9)$ | $2669(11)$ | $2361(14)$ |

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Fig. 1. Mean molecular dimensions ( $\AA$ and deg) $(\sigma=0.01-0.02 \AA$ and $\left.0.6-1.1^{\circ}\right) ; \mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(9)=112, \mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(10)=$ $109^{\circ}$.

Discussion. The geometry and dimensions* of the norbornane skeleton (Fig. 1) are generally normal; the one significant distortion (observed in both molecules) is an increase in the exocyclic angles at $\mathrm{C}(3), \mathrm{C}(4)$, and $\mathrm{C}(5)$, to $114-116^{\circ}$, presumably as a result of steric repulsion between $O(5)$ and $C(9)$ (distance $2.94 \AA$ ).

Molecules are linked in the crystal by a double helical system of hydrogen bonds (Fig. 2).

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[^1]


Fig. 2. Schematic representation of hydrogen-bond system, showing distances $(\AA)$.

## Reference

Allen, M. S., lamb, N., Money, T. \& Salisbury, P. (1979). J. Chem. Soc. Chem. Commun. p. 112.

# Decahydro-5-methylene-2-oxo-6 $\beta$-trimethylsiloxy-[3a $\alpha, 6 \alpha, 8 a \beta]$ ]-3a,6-methano-3 $\alpha$ azulenecarboxylic Acid Methyl Ester 

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#### Abstract

C}_{17} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si}\), monoclinic, $\mathrm{C} 2 / \mathrm{c}, \quad a=$ 35.663 (13), $b=6.608$ (4), $c=15.165$ (5) $\AA, \beta=$ 94.44 (3) ${ }^{\circ}, M_{r}=322 \cdot 48, U=3563 \cdot 1 \AA^{3}, Z=8, D_{m}=$ 1.205 (flotation, aqueous KI), $D_{c}=1.202 \mathrm{Mg} \mathrm{m}^{-3}$, $F(000)=1392$; Mo $K \alpha$ radiation ( $\lambda_{a 1}=0.70926 \AA$, $\mu=0.108 \mathrm{~mm}^{-1}$ ). The structure was solved by direct methods and refined by full-matrix least squares to $R=$ 0.063 using 1614 diffractometer reflexions. The position of the methoxycarbonyl group on the azulenone precursor has been established as $3 \alpha$.

Introduction. The title compound (Fig. 1) was prepared as an intermediate during studies toward the synthesis


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of gibberellic acid (Stork, Still \& Singh, 1979). The structure determination was undertaken to establish the position of the methoxycarbonyl group on the azulenone precursor. The four possibilities were $1 \alpha, 1 \beta$, $3 \alpha$, or $3 \beta$.

The quality of the crystal used for data collection (approximate dimensions $0.2 \times 0.6 \times 0.3 \mathrm{~mm}$ ) was examined on the diffractometer, by taking $\omega$ scans of several strong low-angle reflexions, and judged to be acceptable. The average width of the peaks at half height was $0.23^{\circ}$. Intensity data, in the range $3 \leq 2 \theta \leq$ $55^{\circ}$, were collected at 301 K employing the $\omega-2 \theta$ scan technique on an Enraf-Nonius CAD-4F $\kappa$-geometry © 1979 International Union of Crystallography


[^0]:    * Lists of thermal parameters, calculated hydrogen parameters, molecular dimensions, and structure factors have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34743 ( 19 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH 1 2HU, England.

[^1]:    * See previous footnote.

